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## Trophic transfer of Cd from natural periphyton to the grazing mayfly *Centroptilum triangulifer* in a life cycle test

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Periphyton is a major source of Cd bioaccumulation in a grazing mayfly.

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## ABSTRACT

In streams, periphyton biofilms are important sinks for trace metals such as cadmium and are primary food sources of many invertebrate consumers. To study Cd trophic transfer, we produced differentially contaminated diets by exposing natural periphyton to environmentally relevant dissolved Cd ranging from 0 to 10  $\mu\text{g L}^{-1}$  for 6–7 days using a radiotracer approach. On average, periphyton grown during three different seasons bioconcentrated Cd similarly – approximately 1315 ( $\pm 442$ ) -fold above dissolved concentrations. However, mayfly larvae (*Centroptilum triangulifer*) raised on these differentially contaminated diets (first instar through adulthood) had significantly higher trophic transfer factors from periphyton grown in Aug and Nov 2008 ( $4.30 \pm 1.55$ ) than from periphyton grown in Jan 2009 ( $0.85 \pm 0.21$ ). This Cd bioaccumulation difference is only partially explained by apparent food quality and subsequent growth differences. Taken together, these results suggest that primary producers at the base of food webs drive metal bioaccumulation by invertebrate grazers.

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### 1. Introduction

Insects are the dominant invertebrate faunal group in most lotic systems and play critical ecological roles. It is important that their bioaccumulation of contaminants such as metals is well understood, both in terms of toxicity issues to the insects themselves, and also their contribution to the diets (and thus trophic transfer) to fish and birds. A growing number of studies highlight the importance of diet in the bioaccumulation of metals in aquatic insects (Croisetiere et al., 2006; Croteau and Luoma, 2008; Irving et al., 2003; Martin et al., 2007; Munger and Hare, 1997). That diet is an important route of exposure may help explain the vast discrepancies (e.g. Buchwalter et al., 2007) that exist between observations of insect sensitivities to metals in nature (Cain et al., 2004; Clements, 1991; Clements et al., 2000) and the apparent insensitivities of insects in standard laboratory toxicity tests (Brix et al., 2005) that only use dissolved exposures. Since primary producers often concentrate metals to several orders of magnitude above dissolved concentrations, this exposure pathway is likely highly important in determining responses of insects to metals in nature.

Laboratory attempts to understand trophic transfer of metals often use simplified diets such as a single algal species or a single prey type. While the use of standardized diets may reduce experimental variability, they do not reflect the complex diets that many organisms experience in nature. In streams, many species of insects graze on periphyton – complex biofilms consisting of diatoms, algae, fungi, bacteria, detritus and extracellular matrix – which can be a sink for metals such as Cd (Brooks et al., 2004; Hill et al., 2000; Meylan et al., 2003; Morin et al., 2008a,b; Selby et al., 1985). Therefore, characterizing the bioaccumulation of metals in periphyton is critical to understanding their subsequent transfer to invertebrate grazers.

Here, we used a radiotracer approach to quantify Cd bioaccumulation into natural periphyton communities, which are the natural diets of many mayfly species including *Centroptilum triangulifer* (Sweeney and Vannote, 1984). We exposed periphyton to a range of dissolved Cd concentrations for 6–7 days prior to transferring them to clean water. Larvae of the parthenogenetic mayfly *C. triangulifer* (Ephemeroptera: Baetidae) were raised on these differentially contaminated diets through adulthood. We combined these dietary bioaccumulation studies with dissolved bioaccumulation studies to approximate the relative contribution of Cd in mayfly tissues derived from dissolved and dietary sources. The importance of diet is discussed in the context of setting water quality criteria.

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## 2. Materials and methods

### 2.1. Test animals

The mayfly *Centroptilum triangulifer* (Ephemeroptera: Baetidae) was obtained from culture at the Stroud Water Research Center (Avondale, PA). The same clone (WCC-2) was used for all studies. Originally described as *Cloeon triangulifer* (McDunnough, 1931), this species has been used in studies of temperature and development (Sweeney and Vannote, 1984), chlordane (Sweeney et al., 1993), and aluminum exposure (Tabak and Gibbs, 1991). Given its suitability to laboratory culture, its development as a standard test organism is warranted and ongoing.

### 2.2. Experimental design

#### 2.2.1. Labeling periphyton with cadmium

Acrylic plates (6.5 × 23 × 0.15 cm) were colonized by periphyton by streaming fresh stream water from White Clay Creek, PA (39°51'47"N, 75°47'07"W) continuously for 2–4 weeks in a greenhouse. These periphyton communities were harvested in Aug 2008, Nov 2008 and Jan 2009 for three distinct sets of experiments. Plates were harvested when periphyton reached a thickness of approximately 1–2 mm and express shipped to North Carolina State University. At this stage, periphyton consisted primarily of diatoms with some bluegreen and green algae, along with some naturally-colonizing consumers (predominantly micro- and meiofauna) (Xie et al., 2009). Microbial and fungal constituents were not characterized.

In each study, the colonized plates were placed in 2.0 L glass bottles holding 1.8 L American Society for Testing and Materials (ASTM) artificial soft water (48 mg L<sup>-1</sup> NaHCO<sub>3</sub>, 30 mg L<sup>-1</sup> CaSO<sub>4</sub>·2H<sub>2</sub>O, 30 mg L<sup>-1</sup> MgSO<sub>4</sub>, and 2 mg L<sup>-1</sup> KCl, pH 7.4). The Aug 2008 plates were exposed to dissolved Cd at nominal concentrations of 0, 0.1, 0.5, 2.5 and 10 µg L<sup>-1</sup> with two replicates per treatment. Each Cd treatment received <sup>109</sup>Cd as CdCl<sub>2</sub> (specific activity: 570.2 Bq ng<sup>-1</sup>), with stable CdCl<sub>2</sub> providing the remainder of the ambient Cd in solution. During labeling the cadmium concentration in the water for each replicate was monitored daily and was calculated based on the mean radioactivity of duplicate 1-mL subsamples. As loss of Cd from solution was rapid, appropriate amounts of <sup>109</sup>Cd and CdCl<sub>2</sub> were added to each replicate after 3 days of exposure to maintain designated concentrations. After six days of Cd exposure, the plates were removed from their respective solutions and placed in bottles containing uncontaminated ASTM soft water. After 1 day of equilibration in clean water, duplicate algal samples (approximately 1 cm<sup>2</sup>) were collected from the center of the plate from each bottle, dried, weighed and assayed for radioactivity. Duplicate 2 mL water samples were collected from each bottle as described above. These dissolved exposures generated differentially contaminated diets for the subsequent mayfly tests (see below).

As mayfly larvae did not survive the highest 2 dietary treatments from the Aug 2008 experiments, the Nov 2008 plates were exposed to a lower range of ambient Cd at 0, 0.02, 0.1, 0.5, and 2.5 µg L<sup>-1</sup>. As was the case in the previous experiment, identical amounts of Cd as <sup>109</sup>CdCl<sub>2</sub> (specific activity: 492.1 Bq ng<sup>-1</sup>) were added to each Cd exposure, with stable CdCl<sub>2</sub> providing the remainder of the ambient Cd in solution. Cadmium was replenished after 3 days of Cd exposure as described above, and the duration of exposure for this set of experiments was 7 days. A parallel experiment was conducted exclusively with stable Cd to rule out the potential influence of radioactivity on mayfly performance. Generally poor survivorship was observed in both radioisotope (44%) and stable experiments (32%) prompting a third study. Finally, the Jan 2009 plates were exposed to dissolved Cd: 0, 0.15, 0.625, 2.5 and 10 µg L<sup>-1</sup> for 6 days, with additions of Cd at day 3 as described above. Each concentration had two replicates. The 0.15 µg L<sup>-1</sup> treatment used exclusively <sup>109</sup>CdCl<sub>2</sub> (specific activity: 454.7 Bq ng<sup>-1</sup>), while the other three treatments were prepared by adding the same amount of <sup>109</sup>Cd and appropriate amounts of CdCl<sub>2</sub> to achieve a final concentration of 0.625, 2.5 and 10 µg L<sup>-1</sup> respectively.

#### 2.2.2. Mayfly life cycle exposure to dietary Cd

After the periphyton plates were loaded with Cd, equilibrated in clean water for 24 h and sampled, 4–6 day old larvae were added to each replicate bottle (2 replicates per concentration). Thirty larvae were added to each bottle for the Aug 08 experiments, and 20 larvae were added to each bottle in subsequent experiments. Each bottle was gently aerated, and the light:dark cycle was natural for the given season with ambient light provided by large laboratory windows. Laboratory temperatures ranged from 19 to 22 °C in all experiments. Cadmium water concentrations were monitored twice per week. Cadmium in the periphyton was determined approximately every 10 days as described above. All endpoints are reported for adult mayflies and include Cd bioaccumulation, growth and survival. *Subimago exuvia* and egg masses collected from adults were also assayed for radioactivity.

Subimagos were collected upon emergence (typically ~30 days) and transferred to humid chambers overnight for the final molt to adulthood. Radioactivity in the adults was determined before and after egg delivery. Adults were stimulated to release eggs by wetting the abdomen with ASTM soft water. Individual adults were then placed in a microcentrifuge tube and frozen at -20 °C. Adults were subsequently oven dried at 60 °C for approximately 48 h and the dry weight of each adult was obtained on a Sartorius CP225D microbalance to the nearest 0.01 mg.

### 2.3. Dissolved uptake and efflux

As monitoring of dissolved Cd over the life cycle test revealed some movement of Cd from the periphyton to the surrounding water, we performed dissolved uptake and elimination experiments to allow us to estimate the relative contributions of dissolved Cd to the overall Cd bioaccumulation in the mayflies. Only the highest (217.8 ng L<sup>-1</sup>; Specific activity: 404 Bq ng<sup>-1</sup>) dissolved concentration measured in the Jan 09 experiments was used to determine the uptake and efflux rate constants to ensure adequate radioactivity in the animals after 24 h of exposure. Five mixed-sized larvae (approximately 15–20 days post hatch) used for each of the three replicates were transferred to a Teflon beaker with 80 mL soft water without aeration at the same light regime as the dietary exposure at room temperature. Radioactivity in the larvae was measured at 3, 6, 12, and 24 h after being thoroughly rinsed in clean soft water. Larvae were not fed during exposures. After 24 h of exposure, the animals from each replicate were transferred to plastic bottles with 1 L clean soft water to monitor Cd efflux. Radioactivity in the larvae was determined daily for seven days with water renewed every day.

### 2.4. Acute toxicity test

To assess the acute toxicity of the presumed most sensitive life stage to dissolved Cd, a 48-h acute toxicity test was conducted on first instar larvae (1–2 days post hatch). Seven larvae were exposed to replicate concentrations of 0, 10, 100, 1000, and 10,000 µg L<sup>-1</sup> Cd in a 8.4 cm petri dish at room temperature. Survivorship was evaluated by microscopic examination after 48 h exposure.

### 2.5. Radioactivity measurement

All measurements of radioactivity in water, periphyton, exuvia, *C. triangulifer* adults and eggs were measured using a Perkin-Elmer Wallac Wizard 1480 automatic gamma counter (Shelton, Connecticut, USA) at 16–32 keV. Samples were counted for 3 min. All counting errors were generally <5%. Appropriate corrections were made for radioactive decay and counting efficiency.

### 2.6. Statistical analysis

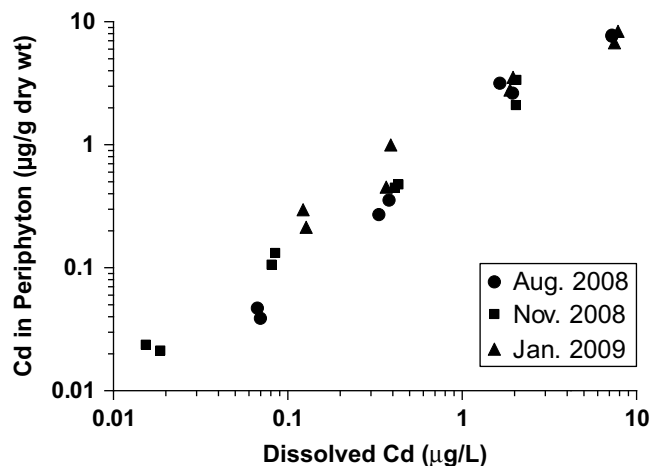
All data are expressed as mean ± standard deviation. Unless otherwise stated, all data analysis was performed using SAS (version 8.02). Bioconcentration of Cd from water to periphyton was determined by dividing Cd concentrations in periphyton (µg g<sup>-1</sup>) by Cd in the water (µg L<sup>-1</sup>), assuming 1 L of soft water weighs 1000 g. Note that these are not steady state bioconcentration factors (BCFs) as they represent periphyton concentrations in a non-equilibrium scenario. Trophic transfer factors (TTFs) were determined by dividing Cd body burden in *C. triangulifer* by mean Cd concentrations in the periphyton over the duration of the experiment after considering dissolved Cd contributions. Cadmium uptake and efflux rate constants were calculated based on models published elsewhere (Schlekat et al., 2002). Cadmium body burden (µg g<sup>-1</sup> wet weight) in the organisms at steady state (C<sub>ss</sub>) from the dissolved route was estimated using the equation: C<sub>ss</sub> = (k<sub>u</sub>/k<sub>e</sub>) \* C<sub>w</sub>, where k<sub>u</sub> is the uptake rate constant (L g<sup>-1</sup> day<sup>-1</sup>), k<sub>e</sub> is the efflux rate constant (day<sup>-1</sup>), and C<sub>w</sub> is Cd concentration in the exposure media (µg L<sup>-1</sup>). Regression analysis and One-way analysis of variance (ANOVA) were applied when appropriate with significance at P < 0.05. LC50 was determined using GraphPad Prism 5 software.

## 3. Results

### 3.1. Cadmium loading of periphyton

The bioconcentration of Cd by periphyton communities was relatively consistent across three separate experiments performed on periphyton grown in three separate seasons (Fig. 1). Average periphyton bioconcentration of Cd from solution was 1315-fold (range 629–2041) higher than ambient dissolved Cd concentrations (Table 1). Periphyton Cd bioconcentration did not show saturation with increasing ambient Cd. We assume that steady state was either reached or approached over the 6–7 days of exposure. The effect of seasonality on the uptake of Cd by periphyton was not statistically significant (P = 0.051) with analysis confounded by differences in exposure duration (6 vs. 7 days).

As each biofilm provided food for a cohort of mayflies for the entire developmental period, we monitored changes in periphyton Cd approximately every 10 days during the mayfly exposures. This schedule allowed us to detect any gross changes in dietary Cd while minimizing the possibility of inadvertently removing larvae during periphyton sampling. Generally, the periphyton maintained

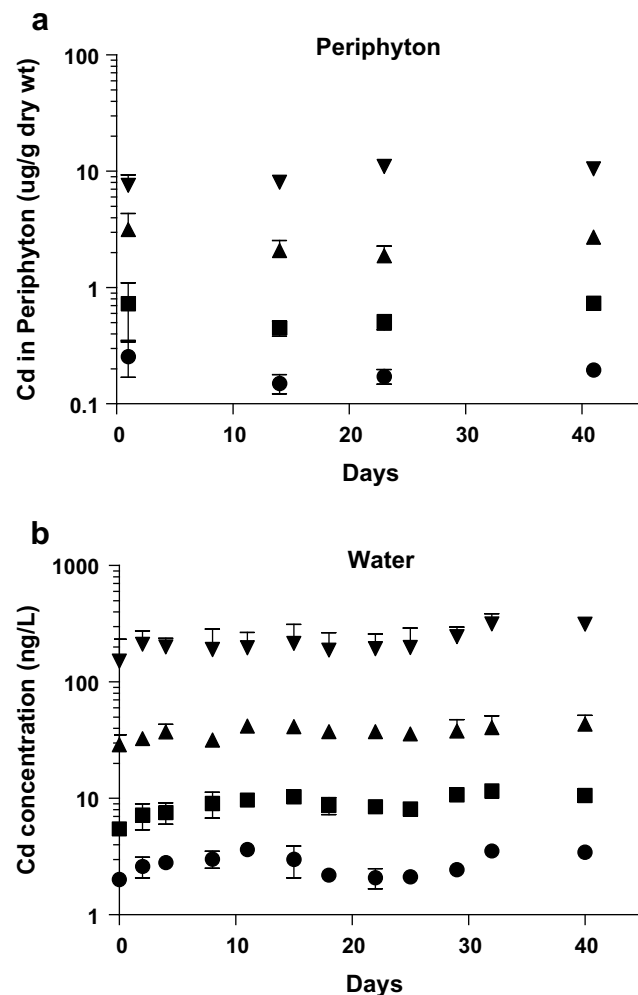


**Fig. 1.** Correlations between dissolved Cd exposure concentration and Cd associated with periphyton biofilms (Pearson  $r > 0.97$  for each of the three experiments). Exposure durations were 6 days (Aug 2008 and Jan 2009 experiments) or 7 days (Nov 2008 experiments), with replenishment of ambient Cd at day 3 in each experiment.

consistent Cd concentrations throughout the mayfly developmental tests (Fig. 2a). Monitoring of Cd in water was more routine (every 2–3 days), and an equilibrium between water and periphyton appeared to be established for each Cd treatment (Fig. 2b).

### 3.2. Life cycle accumulation of cadmium in *C. triangulifer*

Within each experiment, Cd concentrations in *C. triangulifer* adults were strongly correlated with Cd concentrations in the periphyton (Pearson  $r > 0.96$  for all three experiments) (Fig. 3). Despite similar total Cd loading in periphyton across experiments, the accumulation by mayflies was different from experiment to experiment. For example, mayflies grown on periphyton with a Cd concentration of  $0.53 \mu\text{g g}^{-1}$  (Nov 2008 experiment) accumulated approximately 3-fold more Cd than mayflies grown on a slightly higher Cd concentration of  $0.61 \mu\text{g g}^{-1}$  in the Jan 2009 experiments. Trophic transfer factors from the Aug 2008 ( $3.74 \pm 2.60$ ) did not differ from the Nov 2008 ( $4.50 \pm 2.16$ ) experiments but both were larger than those from the Jan 2009 experiments ( $0.85 \pm 0.21$ ) ( $P < 0.01$ ). Residual periphyton in the gut is not an issue in adult



**Fig. 2.** Example of measured Cd concentrations in water (Panel b) and periphyton (Panel a) over the course of *C. triangulifer* life cycle exposure (Jan 2009 experiments). Symbols represent initial nominal dissolved Cd treatments of periphyton ( $\mu\text{g L}^{-1}$ ) – ● – 0.15, ■ – 0.625, ▲ – 0.5, ▼ – 10.0. Error bars represent standard deviations.

**Table 1**

Summary of results from three trophic transfer experiments. Cd in periphyton is reported as the periphyton concentration when mayfly larvae were introduced to differentially contaminated diets. Dissolved Cd concentrations represent the mean of all measured Cd values over the mayfly life cycle exposures. Trophic transfer factors are estimated after accounting for possible dissolved accumulation based on  $k_u$  and  $k_e$  estimates. Percent survival is reported as the mean survival of the 2 replicate exposures. Control survivorship was 98% (Aug. 2008), 50% (Nov. 2008), and 90% (replicate 1), 30% (replicate 2) (Jan. 2009). N/A: not available. All values are reported  $\pm$  standard deviation.

Periphyton treatments (nominal)	Cd in periphyton ( $\mu\text{g g}^{-1}$ dry wt)	Periphyton bioconcentration (fold increase)	Dissolved Cd ( $\text{ng L}^{-1}$ )	Cd in mayfly adults ( $\mu\text{g g}^{-1}$ dry wt)	Trophic transfer factor	% Cd from solution	% Cd from diet	% Survival
<i>Aug 2008</i>								
0.1 $\mu\text{g L}^{-1}$	$0.043 \pm 0.016$	$629 \pm 84$	$6.8 \pm 4.0$	$0.16 \pm 0.17$	3.95	29.5	70.5	57
0.5 $\mu\text{g L}^{-1}$	$0.313 \pm 0.064$	$877 \pm 166$	$16.0 \pm 7.0$	$1.00 \pm 0.74$	3.54	12.5	87.5	40
2.5 $\mu\text{g L}^{-1}$	$2.907 \pm 0.750$	$1617 \pm 212$	$133.5 \pm 58.6$	N/A	N/A	N/A	N/A	0
10 $\mu\text{g L}^{-1}$	$7.722 \pm 2.009$	$1067 \pm 15$	$568.6 \pm 275.7$	N/A	N/A	N/A	N/A	0
<i>Nov 2008</i>								
0.02 $\mu\text{g L}^{-1}$	$0.022 \pm 0.003$	$1326 \pm 103$	$0.8 \pm 0.5$	$0.08 \pm 0.03$	3.38	7.0	93.0	25
0.1 $\mu\text{g L}^{-1}$	$0.119 \pm 0.023$	$1436 \pm 222$	$3.2 \pm 1.1$	$0.84 \pm 0.41$	7.29	3.0	97.0	40
0.5 $\mu\text{g L}^{-1}$	$0.463 \pm 0.206$	$1099 \pm 46$	$11.4 \pm 3.5$	$2.61 \pm 1.31$	4.51	3.6	96.4	50
2.5 $\mu\text{g L}^{-1}$	$2.737 \pm 0.884$	$1344 \pm 438$	$77.2 \pm 28.2$	$9.68 \pm 3.43$	3.08	7.0	93.0	60
<i>Jan 2009</i>								
0.15 $\mu\text{g L}^{-1}$	$0.255 \pm 0.085$	$2041 \pm 466$	$2.7 \pm 0.6$	$0.19 \pm 0.06$	0.87	10.9	89.1	85
0.625 $\mu\text{g L}^{-1}$	$0.726 \pm 0.376$	$1918 \pm 120$	$9.0 \pm 1.9$	$0.66 \pm 0.27$	0.93	10.7	89.3	78
2.5 $\mu\text{g L}^{-1}$	$3.160 \pm 1.170$	$1643 \pm 276$	$37.2 \pm 6.0$	$1.85 \pm 0.78$	0.64	14.9	85.1	65
10 $\mu\text{g L}^{-1}$	$7.568 \pm 1.714$	$787 \pm 150$	$217.8 \pm 72.0$	$11.00 \pm 4.30$	0.95	16.8	83.2	75

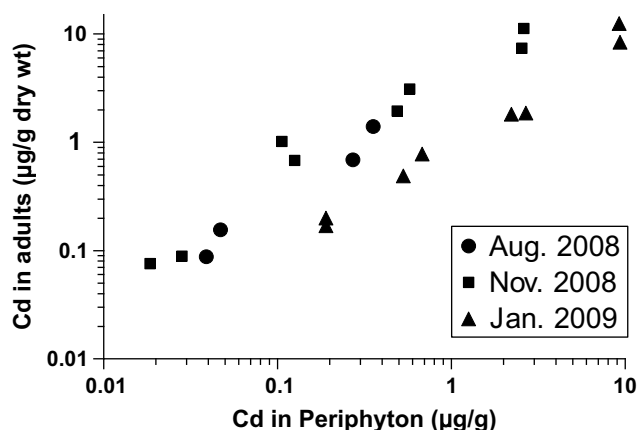


Fig. 3. Relationship between periphyton Cd concentration and Cd body burdens in *C. triangulifer* adults (Pearson  $r > 0.96$  for each of the three experiments).

mayflies, as larvae cease feeding prior to emergence to allow air to fill the gut in the winged life stages (Funk et al., 2008). In all experiments, there was very limited cadmium present in the subimago exuvia and maternal transfer of Cd to eggs was rare (data not shown). A few (<3%) aberrant individuals transferred measurable radioactivity to eggs, but there was no discernable pattern associated with maternal or egg Cd burdens. There was no significant difference in egg production between the control ( $1505 \pm 383$ ,  $n = 6$ ) and combined treatment group adults ( $1373 \pm 288$ ,  $n = 6$ ) ( $P > 0.05$ ).

### 3.3. Dissolved vs. dietary contributions

To estimate the potential contribution of dissolved Cd in mayfly tissues, we estimated the uptake rate ( $k_u = 1.008 \pm 0.26$  ( $L g^{-1} day^{-1}$ )) and efflux rate ( $k_e = 0.041 \pm 0.006 day^{-1}$ ) constants. We applied these constants to the measured dissolved Cd concentrations in each exposure, estimating an assumed steady state Cd body burden derived from solution only. Despite the fact that the  $k_u$  for *C. triangulifer* is the highest reported for an aquatic insect to date (see Buchwalter et al., 2008), dietary Cd was the predominant exposure route. Averaged across all experiments, dietary Cd accounted for 88% of the Cd body burden measured in adult mayfly tissues (Table 1).

### 3.4. Mayfly performance

The survivorship of mayflies (Table 1) was highly variable throughout these studies, perhaps as a result of food quality differences. An apparent concentration–response relationship was observed in the Aug 2008 experiment with excellent control survivorship. Mortality occurred in the highest 2 treatment groups before larvae reached sizes visible to the naked eye. Residual food was present in all treatment groups (excluding controls) in this experiment, suggesting that food availability was not a primary factor in the performance of the Cd treatment groups. A repeat of that experiment (Jan 2009) with similar Cd dietary treatments did not produce any apparent effect on the mayflies. However, as noted above, Cd bioaccumulation was much lower in this experiment. Mayfly survivorship in the Nov 2008 experiments was generally poor in both radioisotope and stable Cd parallel experiments. The average survivorship across all experiments was approximately 60%.

### 3.5. Acute toxicity test

A dissolved acute toxicity assay with newly hatched larvae failed to show toxicity at environmentally relevant concentrations.

The no observed effect concentration (NOEC) of cadmium for the newly hatched *C. triangulifer* larvae was estimated to be  $10 \mu g L^{-1}$ . The 48-h LC50 was determined to be  $352 \mu g L^{-1}$  – a concentration 2–3 orders of magnitude higher than what might be encountered in nature.

## 4. Discussion

Periphyton is a sink for Cd in aquatic environments. In nature, reported Cd concentrations in periphyton vary greatly from different aquatic systems depending on water chemistry and contamination history. For example, in Southeast Missouri streams, Cd in periphyton ranged from  $0.22$  to  $0.43 \mu g g^{-1}$  in streams unaffected by mining activity, and up to  $5.2 \mu g g^{-1}$  (dry wt) in streams affected by lead mining activity (Besser et al., 2007). Periphyton from East Fork Poplar Creek in East Tennessee, a mercury contaminated stream had a Cd level of  $\sim 11 \mu g g^{-1}$  (Hill and Larsen, 2005). Cadmium in periphyton from the Boulder River watershed (Jefferson County, Montana) ranged from  $0.54$  to  $70.3 \mu g g^{-1}$  dry wt (Farag et al., 2007; Rhea et al., 2006). Extreme values ( $143$ – $1809 \mu g g^{-1}$ ) for Cd in periphyton were reported from the Riou Mort River in Southeast France (Morin et al., 2008a,b). Thus, Cd in periphyton from our experiments ( $0.043$ – $7.7 \mu g g^{-1}$ ) is representative of what might be encountered in mildly to moderately contaminated streams.

The high assimilative capacity of periphyton for Cd is also evident from laboratory studies, where exposure to extreme Cd concentrations results in tremendous Cd bioaccumulation in periphyton. For example, exposure to  $64 \mu g L^{-1}$  cadmium for three weeks (Ivorra et al., 2002) and  $25 \mu g L^{-1}$  of cadmium for 10 days (Brooks et al., 2004), resulted in Cd concentrations of approximately 1000 and  $600 \mu g g^{-1}$  dry wt, respectively. This suggests that periphyton is difficult to saturate with Cd under natural conditions and acts as a major Cd sink. In our experiments, periphyton bioconcentrated Cd  $1315 \pm 442$  fold from water. This value is within the range of the reported Cd BCFs of  $8.0 \times 10^2$ – $1.0 \times 10^4$  for freshwater algae (Taylor, 1983), but less than that reported in a marine diatom, which is in the range of  $10^4$ – $10^5$  (Wang and Ke, 2002) and a freshwater diatom *Asterionella formosa* which had a BCF of 40,000 (Conway, 1978).

While periphyton Cd bioaccumulation was comparable among three different seasonal periphyton communities, the subsequent transfer of Cd to grazing mayflies was not. In two of the three experiments, TTFs were quite large (3–4), whereas they were modest ( $< 1$ ) in another. Possible reasons for these bioaccumulation differences among experiments include food quality/growth, and Cd bioavailability differences associated with periphyton flora. Animals grown on the Jan 2009 periphyton were 35% larger on average than animals grown in the other studies ( $P < 0.001$ ). This growth difference was also reflected in the controls of these experiments, suggesting that food quality was superior in the Jan 2009 periphyton.

It is also possible that subcellular Cd compartmentalization varied among the more dominant species in the different periphyton communities. Cadmium is generally adsorbed to the cell wall (insoluble fraction) and absorbed into cytosol (soluble fraction) (Ettajani et al., 2001; Wang and Wang, 2008) in algae. The insoluble fraction is considered to be generally less bioavailable, whereas the cytosolic Cd is considered to be highly bioavailable (Cheung et al., 2006; Ettajani et al., 2001; Rainbow et al., 2007; Wallace and Lopez, 1997; Wang and Wang, 2008). Distributions of Cd can vary widely from species to species (Ettajani et al., 2001; Wang and Wang, 2008). Thus, seasonal differences in periphyton flora may strongly affect Cd bioavailability to grazers. While these compartmentalization differences are tractable in laboratory studies with single

algal species as diets, natural periphyton communities are exceptionally complex. Historical studies of White Clay Creek (Stroud Water Research Center, 1974) report that algal communities are typically comprised of 128–223 species (mean = 168), with the dominant taxa rarely representing more than 15% of the individuals in the algal community.

Our results show that diet is a predominant Cd exposure route and that Cd can be strongly biomagnified from periphyton to *C. triangulifer* in a life cycle assay. Trophic transfer studies for Cd from freshwater primary producers are surprisingly rare, but include TTF estimates of >1 (up to 6) in *Daphnia pulex* (Mathews and Fisher, 2008), and >1 (up to 8) in *Daphnia magna* (Guan and Wang, 2004). In a freshwater food web, Croteau et al. (2005) provide evidence of Cd biomagnification in nature – particularly among invertebrates.

Given that periphyton is a sink for metals, and that metals associated with periphyton can be highly bioavailable, it follows that periphyton diets are a potentially important exposure route for grazing insects in nature. Aside from growth issues associated with dietary exposures (Irving et al., 2003), and potential toxicity observed in the Aug 2008 experiments, the link between bioaccumulation and toxicity remains somewhat elusive. However, the importance of dietary metal bioaccumulation in invertebrates (Hook and Fisher, 2001) calls to question the relevance of toxicity testing that ignores this exposure route. Test systems that incorporate environmentally realistic dietary exposures should be considered in the context of setting water quality criteria.

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